KLH = 12 BSA = 14

KLH = 42 BSA = 43

T = CH3 (31), C2H5 (35)

Q = H (44), CH3 (45), C2H5 (46)

Q = H (49), CH3 (52), C2H5 (55)

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G6PDH
$$\sim$$
NH<sub>2</sub>

Buffer, pH = 7.44

G6PDH $\sim$ NH

G6PDH $\sim$ NH

G6PDH $\sim$ NH

(bromoacetyl-G6PDH)

# FIG. 9

NH<sub>2</sub>

$$(9)$$

$$S \rightarrow 2$$

$$G6PDH^{\bullet\bullet}NH$$

$$(bromoacetyl-G6PDH)$$

$$G_6PDH$$

# FIG. 10

$$\begin{array}{c|c} & & & \\ &$$

FIG. 11